Stereoselective syntheses of a di-, tri-, and tetra-saccharide fragment of *Shigella dysenteriae* type 1 O-antigen using 3,4,6-tri-O-acetyl-2-azido-2-deoxy-α-D-glucopyranosyl chloride as a glycosyl donor *

Viliam Pavliak ¹, Pavol Kováč, and Cornelis P.J. Glaudemans National Institutes of Health, Bethesda, MD 20892 (USA) (Received August 5th, 1991: accepted October 17th, 1991)

ABSTRACT

Methyl 2,4-di-O-benzoyl- α -L-rhamnopyranoside (1) furnished a crystalline 3-O-bromoacetyl derivative that was treated with the dichloromethyl methyl ether–ZnCl₂ reagent to give 2,4-di-O-benzoyl-3-O-bromoacetyl- α -L-rhamnopyranosyl chloride (3). Compounds 1 and 3 were condensed under the conditions of base-deficient, silver trifluoromethanesulfonate-mediated glycosylation to give a fully protected rhamnobioside, which on O-debromoacetylation afforded the disaccharide nucleophile 10. Similar condensation of 3 with methyl 3-O-benzoyl-4,6-O-benzylidene- α -D-galactopyranoside, followed by O-debromoacetylation and condensation of the thus formed methyl O-(2,4-di-O-benzoyl- α -L-rhamnopyranosyl)-(1 \rightarrow 2)-4,6-O-benzylidene-3-O-benzoyl- α -D-galactopyranoside again with 3, gave the trisaccharide glycoside. Subsequent O-debromoacetylation gave 17, having only HO-3³ unsubstituted. Silver perchlorate-mediated glycosylations of 1, 10, and 17 with 3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl chloride afforded, with high α stereoselectivity, protected di-, tri-, and tetra-saccharide glycosides. Subsequent hydrogenation, followed by N-acetylation and O-deacylation, afforded three oligosaccharide glycosides having nonreducing terminal 2-acetamido-2-deoxy- α -D-glucopyranosyl residues and comprising successively larger portions of the repeating unit of *Shigella dysenteriae* type 1 O-antigen.

INTRODUCTION

Lipopolysaccharides (LPSs) are cell-wall components of Gram-negative bacteria and determine their virulence². Shigella dysenteriae type 1 (Shiga's bacillus) provokes severe forms of dysentery, especially in infants and young children. It has been pointed out that the propensity of this disease to pandemic spread, and its resistance to most clinically relevant antibiotics, make a vaccine to prevent Shiga

Correspondence to: Dr. P. Kováč, National Institutes of Health, Bethesda, MD 20892, USA.

^{*} Part 2 of the series Synthesis of ligands related to the O-specific antigen of Shigella dysenteriae type 1. For Part 1, see ref. 1.

¹ Present Address: Division of Biological Sciences, N.R.C., Ottawa, Canada K1A 0R6.

dysentery a high priority³. We wish to study the interaction of antibodies to *S. dyseneriae* type 1 and its antigen, and explore the possibilities of developing a synthetic vaccine against this devastating disease. The O-specific side chain of *Shigella* LPS consists of a variable number of identical oligosaccharide repeating units. It extends away from the cell into the surrounding medium, and determines the immunological specificity of these organisms⁴. Based on chemical analysis of the LPS, Dmitriev et al.⁵ concluded that the chemical repeating unit of the O-antigenic polysaccharide of *S. dysenteriae* type 1 is the tetrasaccharide shown below. From results of genetic and biochemical analysis, Sturm et al.⁶

→ 3)-
$$\alpha$$
-L-Rha p -(1 → 3)- α -L-Rha p -(1 → 2)- α -D-Gal p -(1 → 3)- α -D-Glc p NAc-(1 → D

concluded that the sugar linked to the core is α -D-galactopyranose, and thus that the biochemical repeating unit of the O-antigen is the sequence DABC.

Our study of antigen-antibody interactions involving antibodies specific to the Shigella O-side chain requires a large number of ligands. The synthesis of oligosaccharides related to the sequence ABC has been reported⁷. Also, we have prepared^{8,9} the methyl glycosides (α -D- or α -L-) of a large number of oligosaccharides related to the S. dysenteriae type 1 O-antigen. The present work describes the stepwise synthesis of three oligosaccharidic probes having the sequences DA, DAB, and DABC, respectively. Thus, these ligands all have a 2-acetamido-2-de-oxy- α -D-glucopyranosyl nonreducing end-group. They were obtained as the methyl glycosides.

RESULTS AND DISCUSSION

The stepwise synthesis of the target methyl α -glycosides 8, 15, and 20 required the construction of the α -D-glucosaminyl linkage, most effectively accomplished 10 using glycosyl donors derived from the corresponding 2-azido-2-deoxysugar. According to this strategy, advantage is taken of the nonparticipating nature of the 2-azido group during chemical glycosylation with a glycosyl halide derivative. Thus, the 2-azido-2-deoxy α -glycoside (oligosaccharide) is generated first and, at a desired stage of the overall synthesis, the 2-azido group is converted into the 2-amino or 2-acetamido function. Several new routes to 2-azido-2-deoxy derivatives of p-glucose have recently been described^{1,11-14}. The glycosyl halide we used is 1,3.4,6-tetra-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl chloride (5), first prepared by Paulsen et al.¹⁵ It was previously used in connection with the synthesis of spermidines¹⁶ but, to our knowledge, has never been applied in the synthesis of an oligosaccharide. Because of its lack of reactivity, glycosylations with 5 failed under the standard conditions¹⁷ for coupling in the presence of silver perchlorate as promoter. However, when dichloromethane was used as the solvent, instead of ether¹⁷ or ether-dichloromethane⁸, glycosylations proceeded at a reasonable rate. The 1,2-trans byproducts were not isolated, preventing the direct determination of the degree of steroselectivity. However, high selectivity for the formation of the

1,2-cis linkage was indicated by high yields (70–80%) of the desired α -linked products. Their NMR spectra confirmed the absence of β isomers. It should be noted that these high yields were obtained using only a moderate molar excess (20–40%) of the glycosyl chloride over the acceptor nucleophile. The ¹H-NMR spectra of the products 6, 13, and 18 showed, inter alia, doublets at $\delta \sim 5$ ($J_{1,2} \sim 3.5$ Hz) confirming ¹⁸ that a 1,2-cis glycosidic linkage was formed. A high-field doublet of doublets, characteristic ¹⁸ of the proton in an H-C-N₃ group, was also present. The coupling reactions were carried out without silver carbonate ^{15,19}, eliminating the need for carrying out the reaction in the dark. Thus, the utility of this technique in the synthesis of α -linked N-acetyl-D-glucosamine conjugates, employing the now readily available 5 in preference to the much less accessible β -glycosyl halide ¹⁹, is clearly demonstrated.

To synthesize the methyl α -glycoside of the disaccharide DA, 5 was condensed with methyl 2,4-di-O-benzoyl- α -1-rhamnopyranoside⁸ (1). The product (6) was converted into the corresponding acetamido derivative 7 by catalytic hydrogenolysis over palladium-on-charcoal in the presence of acetic acid and acetic anhydride. Subsequent deacylation gave 8.

We have previously shown²⁰⁻²² that the *O*-bromoacetyl group can be conveniently used as a temporary blocking group in oligosaccharide synthesis. Here, we used this group to advantage in the stepwise construction of the sequence DAB. This synthesis necessitated the preparation of a suitable rhamnobioside (AB) having a free hydroxyl group at C-3², such as 10. Thus, glycoside 1 was treated²⁰ with bromoacetyl bromide in the presence of 1,1,3,3-tetramethylurea, and the crystalline product (2) was cleaved²³ with dichloromethyl methyl ether in the presence of zinc chloride. The glycosyl chloride 3 thus obtained was condensed with 1 under base-deficient conditions with silver trifluoromethanesulfonate (silver triflate) as the promoter. The proton-coupled ¹³C-NMR spectrum of the formed disaccharide (9) contained doublets for C-1² at δ 99.2 ($J_{\text{C-1,H-1}}$ 171.5 Hz) and C-1¹ at 98.4 ($J_{\text{C-1,H-1}}$ 170.8 Hz), characteristic²⁴ of α glycosidic linkages. Compound 9

was O-debromoacetylated with thiourea yielding 10, which was subsequently condensed with 5. The resulting, fully protected trisaccharide was converted into the desired product 15 in analogy with the conversion $6 \rightarrow 8$.

The starting nucleophile in the assembly of the tetrasaccharide DABC was methyl 4,6-O-benzylidene-3-O-benzoyl- α -D-galactopyranoside ²⁵ (4). It was condensed with 3 to give the α -(1 \rightarrow 2)-linked disaccharide glycoside 11. Subsequent O-debromoacetylation of the latter, followed by condensation of the formed 12 with 3, gave the trisaccharide 16, which was converted to the trisaccharide nucleophile 17 by O-debromoacetylation. Finally, condensation of the glycosyl donor 5 with the glycosyl acceptor 17 afforded the tetrasaccharide 18 which, after hydrogenation, N-acetylation, and O-deacylation, afforded the target tetrasaccharide 20. Its proton-coupled ¹³C-NMR spectrum included four doublets showing splittings of 173.0, 170.7, 173.9, and 171.2 Hz, confirming ²⁴ that all glycosidic linkages were α . Features in the ¹³C-(¹H) spectrum, as well as those in the ¹H spectrum (see Experimental) were consistent with the expected structure.

EXPERIMENTAL

General methods.—Unless stated otherwise, optical rotations were measured for solutions in CHCl₃, at 25°, with a Perkin-Elmer automatic polarimeter, Model 241 MC. Thin-layer chromatography (TLC) on precoated slides of Silica Gel G F254

(Analtech) was performed with solvent mixtures of appropriately adjusted polarity consisting of A, toluene-ethyl acetate; B, toluene-acetone; C, hexane-ethyl acetate; D, carbon tetrachloride-acetone; E, chloroform-methanol; and F, hexane-acetone. Detection was effected by charring with 5% H₂SO₄ in ethanol and, when applicable, with UV light. Preparative chromatography was performed by elution from columns of Silica Gel 60 (particle size 0.04-0.063 mm). For chromatography of 3, the silica gel was dried at 140° for 16 h. NMR data were extracted from spectra measured for solutions in CDCl₃ or D₂O, at 25°, with a Varian XL 300 or a Varian Gemini spectrometer. Chemical shifts were measured from the resonance signals of Me₄Si and HDO ($\delta_{HDO} = 4.78$), respectively. For the ¹³C spectra taken in D₂O the internal standard was methanol ($\delta_{MeOH} = 49.0$). Proton signal assignments were made by first-order analysis of the spectra, and were supported by homonuclear decoupling or COSY experiments. Of two magnetically nonequivalent geminal protons, the one resonating at a lower field is denoted Ha and the one at higher field Hb. Carbon signal assignments were made routinely by mutual comparison of the spectra, and by comparison with spectra of related substances⁸. Definitive ¹³C assignments, based on HETCOR experiments, were obtained for compounds 6, 9-12, and 17. The superscripts used for the notation of ring carbon atoms and protons for oligosaccharides, in the text and when reporting the NMR data (Tables I and II, and Experimental), denote the sugar residues or groups containing the designated proton or carbon atom. These are serially numbered, beginning with the residue bearing the aglycon. For example, H-1² refers to H-1 of the second sugar residue, and HO-3³ refers to the hydroxyl group at position 3 of the third sugar residue. Fast-atom bombardment (FAB) mass spectra were obtained with a Jeol SX 102 spectrometer, for samples in a matrix of m-nitrobenzyl alcohol. Reactions requiring anhydrous conditions were performed under argon using common laboratory glassware, and reagents and solvents were handled with Hamilton Series 1000 gas-tight syringes. Unless stated otherwise, solutions in organic solvents were dried with anhydrous sodium sulfate. and concentrated at 2 kPa/40°.

Methyl 2,4-di-O-benzoyl-3-O-bromoacetyl-α-L-rhamnopyranoside (2).—Tetramethylurea (6 mL, 50 mmol), followed by bromoacetyl bromide (6.6 mL, 72.5 mmol), was added at 0° to a solution of methyl 2,4-di-O-benzoyl-α-L-rhamnopyranoside⁸ (9.65 g, 25 mmol) in 1,2-dimethoxyethane (75 mL), and the mixture was stirred at room temperature until TLC (solvent A) showed that the reaction was complete (~3 h). After concentration of the solution the residue was partitioned between H₂O and CH₂Cl₂, and after drying, the organic phase was concentrated to dryness. Crystallization of the residue from EtOH gave pure 2 (11.67 g, 92%); mp 82–83.5°; [α]_D + 77° (c 1.8); ¹³C-NMR (CDCl₃): δ 98.5 (C-1), 71.6 (C-4), 71.2 (C-3), 70.3 (C-2), 66.6 (C-5), 55.3 (OCH₃), 25.0 (COCH₂Br), and 17.7 (C-6); CIMS: m/z 524 [M + 18]⁺.

Anal. Calcd for $C_{23}H_{23}BrO_8$: C, 54.45; H, 4.57; Br, 15.75. Found: C, 53.90; H, 4.50; Br, 15.61.

TABLE I

¹H.NMR data for solutions in CDCl₃ ^a

Com-	Chemical shift	Chemical shifts (8) and multiplicities	licities	***************************************	New regulation profession of the control of the con		- Address - Addr			
punod	H-1	H-2	H-3	H-4	H-5	H-6a	49-Н	ОСН3	CH ₃ CO (CH Ph)	CH_2Br (NH)
2	4.86d	5.48-5.56m ^b	5.64dd	5.48-5.56m ^b	4.10m	1.35d	The second secon			3.63d
n	6.16d	5.71dd	5.91dd	5.591	4.41m	1.39d				3.64d
9	4.87d	5.63dd	4.37dd	5.55t	4.06dt	1.36d		3.46s		
	5.20d	3.23dd	5.05t	4.80t	3.76m	3.85-3.86m ^b	3.85-3.86m b		161s, 1.92s,	
1		,	•	:					2.06s	
_	4.79-4.94m "	5.54dd	4.34dd	5.49t	4.05m	1.36d		3.46s		
	5.01d	4.27dt	4.79-4.94m b	4.79-4.94m b	3.65-3.80m ^b	3.65-3.80m b	3.65-3.80m ^b		1.47s, 1.64s,	
									1.89s, 2.03s	(5,84d)
6	4.90d	5.49dd	4.45dd	5.56t	4.01-4.11m ^b	1.35d		3.468		
	5.14-5.16m ⁶	5.14-5.16m h	5.40dd	5.321	4.01-4.11m ^b	1.16d				3.48d
10	4.89d	5.50dd	4.43dd	5.55t	3.97-4.09m	1.35d		3.468		
	5.17d	5.01dd	3.97-4.09m	5.08t	3.97-4.09m	1.13d				
11	5.09d	4.48dd	5.55dd	4.69dd	3.85s	4.32dd	4.11dd	3.52s	(5.54s)	
	5.24d	5.42dd	5.64dd	5.48t	4.23m	1.32d				3.54d
17	5.11d	4.46dd	5.54dd	4.65 br d	3.84s	4.26-4.33m ^b	4.11bd	3.538	(5.54s)	
	5.24-5.29m ^b	5.24-5.29m ^b	4.26-4.33m b	5.24-5.29m ^b	4.20m	1.31d				
13	4.86 br d	5.50-5.57m ^b	4.44dd	5.50-5.57m ^b	4.01-4.11m ^b	1.36d		3.46s		
	5.14d	5.17 br d	4.01-4.11m ^b	5.36t	4.01-4.11m	1.16d				
	4.66d	3.03dd	4.95t	4.73t	3.531m	3.74dd	~ 3.42dd	3.46s	1.61s, 1.88s,	
									2.01s	

	(2.66d)		3.45d												(2.69d)
	1.34s, 1.62s, 1.85s, 2.00s	(5.54s)		(5.54s)			(5.54s)			1.59s, 1.87s, 2.00s				1.34s, 1.60s,	1.83s, 1.99s
3.46s		3.51s		3.51s			3.52s				3.46s				
	3.36–3.46m ^b	4.09dd		4.10dd			4.11 br d			3.39-3.52m ^b	3.87-3.98m ^b			3.33-3.42m	
1.35 1.17d	3.67dd	4.34dd 1.34d	0.73d	4.31dd	1.33d	0.73d	4.32 br d	1.34d	0.76d	3.71dd	3.87-3.98m ^b	1.32d	0.72d	3.62dd	
3.99-4.13m ^b 3.99-4.13m ^b	3.36-3.46m ^b	3.81-3.86m ^b 4.2m	3.81–3.86m ^b	3.84 br s	4.2m	3.76m	3.76-3.84m ^b	4.21m	3.76-3.84m	3.39-3.52m ^b	3.87-3.98m ^b	4.22m	3.76m	3.33-3.42m h	
5.49-5.55m ^b 5.28t	4.70t	4.66 br d 5.50–5.57m ^b	5.20t	4.67 br d	5.49-5.57m ^b	4.93-5.00m ^b	4.65-4.74m ^b	5.49-5.55m ^b	5.11-5.29m	4.65-4.74m ^b	4.43-4.48m b	5.50t	5.14t	4.821	
4.42dd 3.99-4.13m ^b	4.84t	5.50–5.57m ^b 4.42–4.47m ^b	5.32dd	5.49-5.57m ^b	4.39-4.47m ^b	4.0m	5.49-5.55m ^b	4.40-4.47m ^b	4.00dd	4.92t	5.44dd	$4.33-4.42m^{b}$	3.87-3.98m	4.66t	
5.49–5.55m ^b 5.11 br d	3.99-4.13m ^b		5.09-5.12m ^b	4.39-4.47m ^b	5.38 br dd	4.93-5.00m ^b	$4.40-4.47m^{b}$	5.39 br d	5.11-5.29m	3.02dd	$4.33-4.42m^{b}$	5.33 br s	5.05m ^b	4.06dt	
4.86 br s 5.09 br s	4.47d	5.09-5.12m ^b 5.26d	5.04d	5.11d	5.26 br s	5.08 br s	5.11-5.29m ^b	5.11-5.29m ^b	5.03d	4.65-4.74m ^b	5.05m ^b	5.20 br s	4.96 br s	4.43-4.48m	
7		16		17			81				16				

^a Data in the 1st row of each entry refer to sugar residue 1; data in the 2nd and 3rd rows, if present, refer to sugar residues 2 and 3, respectively. Peak multiplicities: d, doublet; m, multiplet; s, singlet; t, triplet. ^b Overlapping signals.

TABLE II

1H-NMR coupling constants a,b

Compound	Coupling constants (Hz)										
	$\overline{J_{1,2}}$	$J_{2,3}$	$J_{2,\mathrm{NH}}$	J _{3,4}	$J_{4,5}$	J _{5,6a}	J _{5,6b}	$J_{6a,6b}$			
2	1.6	3.4		10.1	c	6.2					
3	1.4	3.4		10.3	9.9	6.2					
6	1.5	3.3		9.9	9.9	6.2					
	3.6	10.5		9.3	9.9	c	c	c			
7	1.9	3.4		10.0	10.0	6.2					
	3.7	10.1	9.8	c	c	c	c	c			
9	1.4	3.4		9.8	9.8	6.2					
	c	3.1		9.3	9.3	6.2					
10	1.5	3.4		9.9	9.9	6.2					
	1.4	3.4		9.8	9.8	6.3					
11	3.4	10.5		3.4	< 1.0	1.2	1.3	12.4			
	1.2	3.3		10.1	9.9	6.2					
12	3.4	10.5		3.4	< 1.0	c	< 1.0	11.7			
	c	c		c	c	6.2					
13	< 1.0	3.4		9.8	c	6.2					
	1.6	3.1		9.8	9.8	6.3					
	3.5	10.5		10.0	10.0	3.0	1.8	12.4			
14	< 1.0	3.4		9.9	9.9	6.2					
	< 1.0	2.9		9.8	9.8	6.9					
	3.5	9.7	9.9	9.7	10.4	2.5	c	12.2			
16	c	c		3.3	< 1.0	1.1	1.1	12.5			
	1.3	3.1		c	c	6.3					
	1.4	3.3		10.0	9.7	6.2					
17	3.4	c		3.4	< 1.0	< 1.0	< 1.0	11.7			
	1.5	3.2		c	9.9	6.2					
	< 1.0	3.2		9.9	9.7	6.2					
18	c	c		c	\boldsymbol{c}	< 1.0	< 1.0	12.3			
	c	c		c	9.9	6.1					
	1.4	3.2		9.8	c	6.2					
	3.6	10.5		9.5	c	2.9	c	12.5			
19	c	10.4		2.7	c	c	c	c			
	< 1.0	c		9.9	9.9	c					
	< 1.0	9.8		9.8	9.8	6.1					
	3.4	9.7	9.9	9.7	9.7	2.1	c	11.9			

^a Data in the 1st row of each entry refer to sugar residue 1; data in the 2nd and 3rd rows, if present, refer to sugar residues 2 and 3, respectively. ^b Measured in CDCl₃. ^c Not determined.

2,4-Di-O-benzoyl-3-O-bromoacetyl- α -L-rhamnopyranosyl chloride (3).—Finely powdered, freshly fused zinc chloride (0.4 g) was added to a solution of the α -glycoside 2 (5.07 g, 10 mmol) in dichloromethyl methyl ether (20 mL), contained in a round-bottomed flask equipped with a dry ice condenser. The mixture was stirred at 70° (bath) with the exclusion of moisture until TLC (solvent B) showed that only traces of the starting material remained (\sim 6 h). The solution was diluted with toluene and concentrated, and the residue was chromatographed to give pure 3 (4.44 g, 87%); $[\alpha]_D + 43^\circ$ (c 1.2); 13 C-NMR (CDCl₃): δ 89.1 (C-1), 72.4, 70.8, 69.8 (2 C), 24.7 (COCH₂Br), and 17.3 (C-6); CIMS: m/z 530 $[M+18]^+$.

Anal. Calcd for C₂₂H₂₀BrClO₇: C, 51.64; H, 3.94; Br, 15.62; Cl, 6.93. Found: C, 51.35; H, 4.10; Br. 15.69; Cl, 6.96.

Methyl O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy-α-D-glucopyranosyl)-(1 \rightarrow 3)-2,4-di-O-benzoyl-α-L-rhamnopyranoside (6).—A solution of 1 (ref. 8, 0.39 g, 1 mmol), 5 (0.49 g, 1.4 mmol), and 2,4,6-trimethylpyridine (0.186 mL, 1.4 mmol) in CH₂Cl₂ (15 mL) was added slowly at room temperature to a stirred suspension of silver perchlorate²⁶ (0.35 g, 1.7 mmol) in CH₂Cl₂ (15 mL). After 45 min, TLC (solvent C) showed almost complete conversion of starting materials into a major product. The mixture was filtered, the filtrate was washed with aq sodium thiosulfate, the organic phase was dried and concentrated, and the residue was chromatographed (solvent A) to give pure 6 (0.505 g, 72%); mp 134–135° (from EtOH), [α]_D + 186° (c 1.1); ¹³C-NMR (CDCl₃): δ 98.8 (C-1¹), 93.7 (C-1²), 72.2 (C-4¹), 71.6 (C-3¹), 70.2 (C-3²), 68.0 (C-4²), 67.4 (C-2¹,5²), 66.4 (C-5¹), 61.3 (C-6²), 60.5 (C-2²), 55.3 (OCH₃), 20.7, 20.5, 20.2 (3 × COCH₃), and 17.6 (C-6¹); CIMS: m/z 717 [M + 18]⁺. Anal. Calcd for C₃₃H₃₇N₃O₁₄: C, 56.65; H, 5.33; N, 6.01. Found: C, 56.57; H, 5.33; N, 6.03.

Methyl O-(2-acetamido-2-deoxy-3,4,6-tri-O-acetyl-α-D-glucopyranosyl)-(1 → 3)-2,4-di-O-benzoyl-α-L-rhamnopyranoside (7).—A solution of **6** (1.28 g, 1.89 mmol) and acetic anhydride (1 mL) in MeOH (100 mL) was stirred under H₂, at room temperature, with 5% Pd–C catalyst (1.3 g) for 24 h. After filtration and evaporation of the solvents, the residue was chromatographed (solvent *D*) to give **7** (1.03 g, 76%); [α]_D + 129° (*c* 1.1); ¹³C-NMR (CDCl₃): δ 98.7 (C-1¹), 95.3 (C-1²), 73.0 (C-4¹), 72.5 (C-3¹), 70.9 (C-3²), 68.5 (C-4²), 67.8 (C-5²), 67.5 (C-2¹), 66.5 (C-5)¹, 61.2 (C-6²), 55.3 (OCH₃), 51.2 (C-2²), 22.3 (NCOCH₃), 20.6, 20.5, 20.2 (3 × COCH₃), and 17.6 (C-6¹); CIMS: m/z 716 [M + 1]⁺.

Anal. Calcd for $C_{35}H_{41}NO_{15}$: C, 58.74; H, 5.77; N, 1.96. Found: C, 58.66; H, 5.80; N, 1.99.

Methyl O-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-(1 \rightarrow 3)-α-L-rhamnopyranoside (8)—A solution of 7 (0.99 g, 1.4 mmol) in MeOH (100 mL) was treated with M methanolic sodium methoxide (1 mL), and the solution was kept overnight at \sim 50° (bath). After cooling to room temperature, neutralization with Amberlite IR-120 (H⁺) resin, and evaporation of the solvent, the crude product was chromatographed (solvent E) to give pure (TLC, solvent E, 3:1, NMR) 8 (0.51 g, 96%) as a white, amorphous, hygroscopic solid; $[\alpha]_D + 98^\circ$ (c 1.1, H₂O); CIMS: m/z 382 [M + 1]⁺. Definite, diagnostically significant signals in the ¹H-NMR spectrum (D₂O) were at δ 4.99 (d, 1 H, $J_{1,2}$ 3.6 Hz, H-1²), 4.69 (d, 1 H, $J_{1,2}$ 1.2 Hz, H-1¹), 4.03 (br d, 1 H, H-2¹), 3.38 (s, 3 H, OCH₃), 2.03 (s, 3 H, NCOCH₃), and 1.31 (d, 3 H, $J_{5,6}$ 6.2 Hz, H-6¹); ¹³C-NMR (CDCl₃): δ 100.8 (C-1¹), 94.4 (C-1²), 75.5 (C-3¹), 71.9, 71.0, 70.4, 69.9, 68.6, 66.8, 60.4 (C-6²), 54.9 (OCH₃), 53.7 (C-2²), 22.0 (NCOCH₃), and 16.8 (C-6¹).

Methyl O-(2,4,di-O-benzoyl-3-O-bromoacetyl- α -L-rhamnopyranosyl)- $(1 \rightarrow 3)$ -2,4-di-O-benzoyl- α -L-rhamnopyranoside (9).—A solution of 1 (1.69 g, 4.4 mmol), 3 (2.75 g, 5.3 mmol), and 2,4,6-trimethylpyridine (0.5 mL, 3.94 mmol) in CH₂Cl₂ (20 mL)

was added dropwise at 22° to a stirred suspension of silver triflate (1.352 g, 5.3 mmol) in CH₂Cl₂ (20 mL). When TLC (solvent *B*) showed complete conversion of the starting materials (~ 1 h) the reaction mixture was worked up as described for the preparation of **6**, and chromatography gave **9** (2.28 g, 60%, white foam); $[\alpha]_D + 112^\circ$ (c 1.7); ¹³C-NMR (CDCl₃): δ 99.2 (C-1², $J_{C,H}$ 171.5 Hz), 98.4 (C-1¹, $J_{C,H}$ 170.8 Hz), 76.0 (C-3¹), 73.3 (C-4¹), 72.2 (C-2¹), 71.3 (C-4²), 70.5 (C-3²), 70.2 (C-2²), 67.5 (C-5²), 66.6 (C-5¹), 55.2 (OCH₃), 24.9 (COCH₂Br), 17.7 (C-6¹), and 17.3 (C-6²); CIMS: m/z 878 [M + 18]⁺.

Anal. Calcd for $C_{43}H_{41}BrO_{14}$: C, 59.94; H, 4.80; Br, 9.27. Found: C, 59.79; H, 4.83; Br, 9.37.

Methyl O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 \rightarrow 3)-2,4-di-O-benzoyl-α-L-rhamnopyranoside (10).—A solution of thiourea (0.66 g, 8.6 mmol) in MeOH (10 mL) was added to a solution of 9 (1.86 g, 2.16 mmol) in 1:1 MeOH-CH₂Cl₂ (20 mL). The mixture was kept at room temperature until TLC (solvent A) showed that the reaction was complete (\sim 30 min), and concentrated. A solution of the residue in CH₂Cl₂ was successively washed with 10% HCl, a mixture of satd aq NaHCO₃ and NaCl, and H₂O, dried, and concentrated. Chromatography gave 10 (1.57 g, \sim 100%); [α]_D + 108° (c 1.4); ¹³C-NMR (CDCl₃): δ 99.3 (C-1²), 98.4 (C-1¹), 76.4 (C-3¹), 75.2 (C-4²), 73.2 (C-4¹), 73.0 (C-2²), 72.4 (C-2¹), 68.5 (C-5²), 67.0 (C-3²), 66.6 (C-5¹), 55.2 (OCH₃), 17.7 (C-6¹), and 17.3 (C-6²); CIMS: m/z 758 [M + 18]⁺.

Anal. Calcd for C₄₁H₄₀O₁₃:C, 66.48; H, 5.44. Found: C, 66.38; H, 5.45.

Methyl O-(2,4-di-O-benzoyl-3-O-bromoacetyl-α-L-rhamnopyranosyl)-(1 → 2)-4,6-O-benzylidene-3-O-benzoyl-α-D-galactopyranoside (11).—A solution of methyl 4,6-O-benzylidene-3-O-benzoyl-α-D-galactopyranoside 25 (0.38 g, 1 mmol), 3 (0.51 g, 1 mmol), and 2,4,6-trimethylpyridine (0.12 mL, 0.9 mmol) in CH₂Cl₂ (5 mL) was treated with silver triflate (0.31 g, 1.2 mmol) as described for the preparation of 9. After workup, as described above, crystallization from EtOH gave 0.65 g (75%) of 11; mp 160–162°; [α]_D + 144° (c 1.8); 13 C-NMR (CDCl₃): δ 100.7 (CH-benzylic), 99.9 (C-1¹), 99.6 (C-1²), 75.2 (C-2¹), 74.3 (C-4¹), 71.7 (C-4²), 70.8 (C-3²), 70.7 (C-3¹), 70.4 (C-2²), 69.2 (C-6¹), 67.3 (C-5²), 62.3 (C-5¹), 55.6 (OCH₃), 24.9 (COCH₃), and 17.7 (C-6²); CIMS: m/z 878 [M + 18]⁺.

Anal. Calc. for C₄₃H₄₁BrO₁₄: C, 59.94; H, 4.80; Br, 9.27. Found: C, 59.86; H, 4.83; Br, 9.34.

Methyl O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 \rightarrow 2)-4,6-O-benzylidene-3-O-benzoyl-α-D-galactopyranoside (12).—To a solution of 11 (4.4 g, 5.1 mmol) in 1:1 MeOH-CH₂Cl₂ (70 mL) was added thiourea (1.55 g, 20.4 mmol) and 2,4,6-trimethylpyridine (0.62 g, 5.1 mmol) in MeOH (35 mL), and the mixture was stirred at room temperature until TLC (solvent A) showed that the reaction was complete (\sim 30 min). The mixture was processed as described for the preparation of 10, and chromatography of the crude product gave 12 (3.7 g, \sim 100%); [α]_D + 124° (α 1.0); α 13°C-NMR (CDCl₃): α 100.6 (α 10.6 (α 1.0); 99.8 (α 1.1), 99.6 (α 1.1), 75.3, 73.1

 $(C-2^2, 4^2)$, 75.2 $(C-2^1)$, 74.3 $(C-4^1)$, 70.3 $(C-3^1)$, 69.1 $(C-6^1)$, 68.5 $(C-3^2)$, 66.8 $(C-5^2)$, 62.1 $(C-5^1)$, 55.5 (OCH_3) , and 17.7 $(C-6^2)$; CIMS: m/z 758 $[M+18]^+$.

Anal. Calcd for C₄₁H₄₀O₁₃: C, 66.48; H, 5.44. Found: C, 66.33; H, 5.47.

Methyl O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy-α-D-glucopyranosyl)-(1 → 3)-O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 → 3)-2,4-di-O-benzoyl-α-L-rhamnopyranoside (13).—A solution of 10 (1.48 g, 2 mmol), 5 (0.98 g, 2.8 mmol), and 2,4,6-trimethyl-pyridine (0.37 mL, 2.8 mmol) in CH₂Cl₂ (10 mL) was added at 0° to a suspension of silver perchlorate (0.7 g, 3.4 mmol) in CH₂Cl₂ (30 mL), and the mixture was stirred until TLC (solvent *A*) showed complete conversion of the starting materials (~ 1 h). One major product was formed. Workup as described for the preparation of 6 and chromatography (solvent *C*) gave 13 (1.74 g, 82%); [α]_D + 177° (*c* 0.95); ¹³C-NMR (CDCl₃): δ 99.3 (C-1²), 98.5 (C-1¹), 93.9 (C-1³), 76.1 (C-3¹), 73.3 (C-4¹), 72.0, 71.9, 71.8 (C-2², 3², 4²), 70.3 (C-3³), 67.8, 67.6, 67.3, 67.2 (C-2¹, 5², 4³, 5³), 66.6 (C-5¹), 60.9 (C-6³), 60.4 (C-2³), 55.3 (OCH₃), 20.6, 20.5, 20.2 (3 × OCOCH₃), 17.7 (C-6¹), and 17.4 (C-6²); CIMS: *m/z* 1071 [M + 18]⁺.

Anal. Calcd for $C_{53}H_{55}N_3O_{20}$: C, 60.40; H, 5.26; N, 3.99. Found: C, 60.37; H, 5.30; N, 3.97.

Methyl O-(2-acetamido-2-deoxy-3,4,6-tri-O-acetyl-α-D-glucopyranosyl)-(1 → 3)-O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 → 3)-2,4-di-O-benzoyl-α-L-rhamnopyranoside (14).—A solution of 13 (1.71 g, 1.62 mmol) in MeOH (100 mL) was hydrogenated in the presence of acetic anhydride (1 mL), as described for the preparation of 7. After processing as for 7 and chromatography (solvent *B*), amorphous 14 (1.2 g, 70%) showed $[\alpha]_D$ + 157° (*c* 1.2); ¹³C-NMR (CDCl₃): δ 99.2 (C-1²), 98.5 (C-1¹), 95.7 (C-1³), 76.1 (C-3¹), 73.3, 73.1 (C-4¹, 4²), 72.2, 72.0 (C-2², 3²), 71.0 (C-3³), 68.7 (C-4³), 67.7, 67.4, 67.2 (C-2¹, 5², 5³), 66.5 (C-5¹), 60.9 (C-6³), 55.3 (OCH₃), 50.9 (C-2³), 22.2 (NCOCH₃), 20.5, 20.2 (1 C, 2 C, 3 × COCH₃), 17.7 (C-6¹), and 17.4 (C-6²); CIMS: m/z 1070 [M + 1]⁺.

Anal. Calc. for $C_{55}H_{59}NO_{21}$: C, 61.73; H, 5.56; N, 1.31. Found: C, 61.80; H, 5.58; N, 1.29.

Methyl O-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-(1 → 3)-O-α-L-rhamnopyranosyl-(1 → 3)-α-L-rhamnopyranoside (15).—Deprotection of 14 (1 g, 0.93 mmol), as described for the preparation of compound 8, and chromatography (solvent E) gave pure (TLC, solvent E, 2:1; NMR) 15 (0.38 g, 77%) as a white, amorphous, hygroscopic solid, $[\alpha]_D + 38^\circ$ (c 1, H₂O). Definite, diagnostically significant signals in the ¹H-NMR spectrum (D₂O) were at δ 5.03 (br d, 2 H, overlapping H-1²,1³), 4.65 (br s, H-1¹), 3.38 (s, 3 H, OCH₃), 2.03 (s, 3 H, NCOCH₃), 1.29, 1.28 (2 d, J 6.2 and J 6.1 Hz, H-6¹, 6²); ¹³C-NMR (D₂O): δ 102.1 (C-1²), 100.9 (C-1¹), 94.4 (C-1³), 78.1 (C-3¹), 75.5 (C-3²), 72.0, 71.5, 71.0, 70.4, 69.9 (2 C), 69.2, 68.7, 67.0, 60.5 (C-6³), 54.9 (OCH³), 53.7 (C-2³), 22.0 (NCOCH₃), 16.8, and 16.7 (C-6¹, 6²); CIMS: m/z 528 [M + 1]⁺.

Methyl O-(2,4-di-O-benzoyl-3-O-bromoacetyl- α -L-rhamnopyranosyl)-(1 \rightarrow 3)-O-(2,4-di-O-benzoyl- α -L-rhamnopyranosyl)-(1 \rightarrow 2)-4,6-O-benzylidene-3-O-benzoyl- α -D-galactopyranoside (16).—A solution of 3 (1.18 g, 2.3 mmol), 12 (1.48 g, 2 mmol),

and 2,4,6-trimethylpyridine (0.24 mL, 1.8 mmol) in CH₂Cl₂ (15 mL) was allowed to react with silver triflate (0.62 g, 2.4 mmol) as described for the preparation of 9. The mixture was worked up as for 9 and chromatographed (solvent C) to give 16, and also some unchanged 12. Yield, 1.67 g (70%, or 92% based on the amount of nucleophile consumed); $[\alpha]_D + 129^\circ$ (c 1.1); 13 C-NMR (CDCl₃): δ 100.7 (CH-benzylic), 99.8 (C-1¹), 99.7 (C-1²), 99.0 (C-1³), 75.8 (C-3²), 75.5 (C-2¹), 74.4 (C-4¹), 73.1 (C-4²), 72.2 (C-2²), 71.2 (C-4³), 70.4 (C-3³), 70.3 (C-3¹), 70.1 (C-2³), 69.2 (C-6¹), 67.4 (C-5²), 67.2 (C-5³), 62.2 (C-5¹), 55.5 (OCH₃), 24.9 (COCH₂Br), 17.8 (C-6²), and 16.8 (C-6³); CIMS: m/z 1234 [M + 18]⁺.

Anal. Calcd for $C_{63}H_{59}BrO_{20}$: C, 62.22; H, 4.89; Br, 6.57. Found: C, 62.65; H, 5.09; Br, 6.65.

Methyl O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 → 3)-O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 → 2)-4,6-O-benzylidene-3-O-benzoyl-α-D-galactopyranoside (17).—Compound 16 (1.31 g, 1.1 mmol) was O-debromoacetylated with thiourea (0.33 g, 4.3 mmol) in the presence 2,4,6-trimethylpyridine (0.14 mL, 1.1 mmol), as described for the preparation of 12. Chromatography of the crude product (solvent B) gave 17 (1.03 g, 87%); $[\alpha]_D + 124^\circ$ (c 0.9); 13 C-NMR (CDCl₃): δ 100.7 (CH-benzylic), 99.8 (C-1¹), 99.6 (C-1²), 99.1 (C-1³), 75.9 (C-3²), 75.8 (C-2¹), 75.1 (C-4³), 74.4 (C-4¹), 73.0 (C-4²), 72.8 (C-2³) 72.4 (C-2²), 70.3 (C-3¹), 69.2 (C-6¹), 68.4 (C-3³), 67.4 (C-5²), 66.8 (C-5³), 62.2 (C-5¹), 55.5 (OCH₃), 17.8 (C-6²), and 16.9 (C-6³); CIMS: m/z 1112 [M + 18]⁺.

Anal. Calcd for C₆₁H₅₈O₁₉: C, 66.90; H, 5.34. Found: C, 66.71; H, 5.29.

Methyl O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy-α-D-glucopyranosyl)-(1 \rightarrow 3)-O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 \rightarrow 3)-O-(2,4-di-O-benzoyl-α-L-rhamnopyranosyl)-(1 \rightarrow 2)-4,6-O-benzylidene-3-O-benzoyl-α-D-galactopyranoside (18).—A solution of 17 (1.87 g, 1.7 mmol), 5 (0.83 g, 2.4 mmol), and 2,4,6-trimethylpyridine (0.32 mL, 2.4 mmol) in CH₂Cl₂ (10 mL) was treated with silver perchlorate (0.6 g, 2.9 mmol) as described for the preparation of 13. After workup as for 13 and chromatography (solvent *F*) the amorphous 18 (1.72 g, 72%) showed [α]_D + 191° (α 1.1); ¹³C-NMR (CDCl₃): δ 100.7 (α 1.1); 13C-NMR (C-3²), 75.5 (C-2¹), 74.4, 73.1, 72.1, 71.8, 71.6, 70.3 (2 C), 69.2 (C-6¹), 67.7, 67.4, 67.2, 62.2 (C-5¹), 60.9 (C-4⁴), 60.4 (C-2⁴), 55.5 (α 1.2); 20.5, 20.4, 20.2 (3 × COCH₃), 17.8 (C-6²), and 16.7 (C-6³); CIMS: α 1430 [M + H]⁺.

Anal. Calcd for $C_{73}H_{73}N_3O_{26}$: C, 62.26; H, 5.23; N, 2.99. Found: C, 62.49; H, 5.32; N, 2.96.

Methyl O-(2-acetamido-2-deoxy-3,4,6-tri-O-acetyl- α -D-glucopyranosyl)-(1 \rightarrow 3)-O-(2,4-di-O-benzoyl- α -L-rhamnopyranosyl)-(1 \rightarrow 3)-O-(2,4-di-O-benzoyl- α -L-rhamnopyranosyl)-(1 \rightarrow 2)-3-O-benzoyl- α -D-galactopyranoside (19).—A solution of 18 (1.65 g, 1.17 mmol) and acetic anhydride (1 mL) in MeOH (100 mL) was stirred in a H₂ atmosphere with 5% Pd-C catalyst (1.65 g) for 24 h. TLC (solvent B) showed complete conversion of 18 into two products, both showing lower mobility than the starting material. Acetic acid (20 mL) was added, and hydrogenation was continued for another 24 h, when TLC showed that the faster moving product was almost

fully converted into the slower moving one. After processing as described for the preparation of **8**, and concentration and coevaporation with toluene to remove acetic acid, chromatography of the residue gave **19** (0.78 g, 50%); $[\alpha]_D + 169^\circ$ (c 1.4); 13 C-NMR (CDCl₃): δ 100.0 (C-1¹), 99.4 (C-1²), 99.0 (C-1³), 95.7 (C-1⁴), 76.4 (C-3²), 75.7 (C-2¹), 73.1, 73.0, 72.2 (2 C), 72.0, 70.1, 69.7, 68.9, 68.5, 67.6, 67.3, 67.2 (2 C), 63.2 (C-6¹), 60.9 (C-6⁴), 55.3 (OCH₃), 50.9 (C-2⁴), 22.1 (NCOCH₃), 20.5 (2 C), 20.2 (3 × COCH₃), 17.8 (C-6²), and 16.8 (C-6³); CIMS: m/z 1336 [M + 1]⁺. Anal. Calcd for C₆₈H₇₃NO₂₇: C, 61.12; H, 5.51; N, 1.05. Found: C, 61.21; H, 5.51; N, 1.04.

Methyl O-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-(1 → 3)-O-α-L-rhamnopyranosyl-(1 → 3)-O-α-L-rhamnopyranosyl-(1 → 2)-α-D-galactopyranoside (20).— Deprotection of 19 (0.57 g, 0.43 mmol) as described for the preparation of compound 8, and chromatography (solvent *E*) of the crude product, gave pure (TLC, solvent *E*, 1.5:1; NMR) 20 (0.19 g, 65%) as a white, amorphous, hygroscopic solid; $[\alpha]_D$ + 75° (*c* 1, H₂O). Definite, diagnostically significant signals in the ¹H-NMR spectrum (D₂O) were at δ 5.04 (br d, 2 H overlapping, H-1³, 1⁴), 4.91 (d, 1 H, J_{1,2} 3.7 Hz, H-1¹), 4.89-(br s, 1 H, H-1²), 4.18, 4.13 (2 br s, 2 H, H-2², 2³), 3.42 (s, 3 H, OCH₃), 2.03 (s, 3 H, NCOCH₃), 1.30 (d, 6 H, J_{5,6} 6.1 Hz, H-6², 6³); ¹³C-NMR (D₂O): δ 102.9, 102.4 (2 C₂ ¹J_{C,H} 173.0 and 170.7 Hz, resp., C-1², 1³), 99.1 (¹J_{C,H} 173.9 Hz, C-1¹), 94.6 (¹J_{C,H} 170.7 Hz, C-1⁴), 78.8 (C-3²), 77.6 (C-2¹), 75.6 (C-3³), 72.1, 71.4, 71.2, 70.9, 70.6, 70.0 (2 C), 69.6 (2 C), 69.5, 68.7, 67.1, 61.5 (C-6¹), 60.6 (C-6⁴), 55.1 (OCH₃), 53.9 (C-2⁴), 22.2 (NCOCH₃), 17.1, and 17.0 (C-6², 6³); CIMS: *m/z* 690 [M + 1]⁺.

REFERENCES

- 1 V. Pavliak and P. Kováč, Carbohvdr, Res., 210 (1991) 333-337.
- 2 H. Watanabe and K.N. Timmis, Infect. Immun., 43 (1984) 391-396.
- 3 M.M. Levine, Lancet, 335 (1990) 958-961.
- 4 M.W. Binns, M.S. Vaughan, and K.N. Timmis, Zentralbl. Bakteriol., Mikrobiol. Hyg., Abt. 1, Orig. B, 181 (1985) 197-205.
- 5 B.A. Dmitriev, Yu.A. Knirel, and N.K. Kochetkov, Eur. J. Biochem., 66 (1976) 559-566.
- 6 S. Sturm, B. Jann, K. Jann, P. Fortnagel, and K.N. Timmis, Microb. Pathog., 1 (1986) 307-324.
- 7 B. Classon, P.J. Garegg, and C. Hällgren, Abstr. XIVth Int. Carbohydr. Symp., August 1988, Stockholm, Sweden, p. 245.
- 8 P. Kováč and K.J. Edgar, J. Org. Chem., 57 (1992) in press.
- 9 V. Pozsgay, H. Yeh, C.P.J. Glaudemans, C.-Y. Chy, R. Schneerson, and J.B. Robbins, unpublished.
- 10 H. Paulsen, Angew. Chem. Int. Ed. Engl., 21 (1982) 155-173.
- 11 J.N. Vos, J.H. van Boom, C.A.A. van Boeckel, and T. Beetz, J. Carbohydr. Chem., 3 (1984) 117-124.
- 12 M. Kloosterman, M.P. de Nijs, and J.H. van Boom, J. Carbohydr. Chem., 5 (1986) 215-233.
- 13 F. Dasgupta and P.J. Garegg, J. Chem. Soc., Chem. Commun., (1989) 1640-1641.
- 14 F. Dasgupta and P.J. Garegg, Synthesis, (1988) 626-629.
- 15 H. Paulsen, A. Richter, V. Sinnwell, and W. Stenzel, Carbohydr. Res., 64 (1978) 339-364,
- 16 Y. Saito, T. Watanabe, H. Hashimoto, and J. Yoshimura, Carbohydr. Res., 169 (1987) 171-188.
- 17 K. Igarashi, J. Irisawa, and T. Honma, Carbohydr. Res., 39 (1975) 213-225.
- 18 H. Paulsen and W. Stenzel, Ber., 111 (1978) 2334-2347.
- 19 H. Paulsen and W. Stenzel, Ber., 111 (1978) 2348-2357.
- 20 P. Kováč and R.B. Taylor, Carbohydr. Res., 167 (1987) 153-173.

- 21 T. Ziegler, P. Kováč, and C.P.J. Glaudemans, Carbohydr. Res., 194 (1989) 185-198.
- 22 T. Ziegler, P. Kováč, and C.P.J. Glaudemans, Carbohydr. Res., 203 (1990) 253-263.
- 23 H. Gross, I. Farkas, and R. Bognár, Z. Chem., 18 (1978) 201-210.
- 24 K. Bock and C. Pedersen, J. Chem. Soc., Perkin Trans. 2, (1974) 293-297.
- 25 N. Dang, V.R.N. Munasinghe, and W.G. Overend, J. Chem. Soc., Perkin Trans. 1, (1983) 257-363.
- 26 P. Kováč and L. Lerner, Carbohydr. Res., 184 (1988) 87-112.